

CHF enhancement in flow boiling system with TSP and boric acid solutions under atmospheric pressure

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1. Introduction

In several studies, it was demonstrated that surfactant affects the heat transfer with the change of fluid property related with bubble formation and behavior due to the decrease of surface tension. G. Hetsroni et al. [1] showed in their experiments with Habon G surfactant that the heat transfer of the boiling process is enhanced considerably by the addition of a small amount of surfactant. It was also experimented by Rozenblit et al. [2] on how flow changes in surfactant solution. It showed that gas bubbles in air-water solution with surfactant are smaller in size but much larger in number than in pure air-water mixture, at all flow regimes. In recent, an experiment was performed by Jeong et al. [4], which was on flow boiling CHF enhancement with TSP surfactant solutions under atmospheric pressure in SS316 circular tube. It was represented that the addition of TSP surfactant helps to increase wettability by reducing surface tension and the increment is vary with the concentration of TSP, mass flux, and subcooling.

Tri-sodium phosphate (TSP, $\text{Na}_3\text{PO}_4 \cdot 12\text{H}_2\text{O}$) and boric acid (H_3BO_3) are used in nuclear power plant for maintaining high pH level during accidents. It is noticed that boric acid concentration is about 2000 ppm at normal condition and about 4000 ppm in IRWST.

In this study, the variation of the CHF is observed in water with TSP or boric acid at circular tube of flow boiling water system under atmospheric pressure. Same procedure of the experiment is performed on several concentrations of TSP or boric acid, and it was shown that TSP and boric acid are effective for the CHF enhancement from the results of the experiment.

2. Methods and Results

2.1. Experimental Works

The flow boiling CHF experiment was performed using the low pressure water CHF test loop at the Korea Advanced Institute of Science and Technology (KAIST). A test section was directly heated using an electrical DC power supply unit.

Thermocouples, pressure transducer, flow transducer, and other instruments were connected to a HP 3852A data acquisition/control unit for data collection and processing.

The test section is a circular tube of SS316 installed vertically in the test loop. In the test section, fluid flows in vertical upward direction. The temperatures of inlet and outlet of the test section were measured by in-stream T-type sheathed thermocouples, and the pressures of inlet and outlet of the test section were measured by the pressure transducers. Five K-type thermocouples (outer diameter is 1.5 mm) were installed at the outer surface of the test section to measure the wall temperature and detect onset of the CHF. The first thermocouple was attached at 5 mm below the top of the heated tube, and the rest of the thermocouples are installed at 10 mm intervals.

The experiment was performed in flow boiling system under atmospheric pressure. Mass flux ranged from 100 to 500 $\text{kg/m}^2\text{s}$ in the experiment. And two inlet temperatures (50 °C) were used. The test condition is shown in Table 1.

Table 1. Test matrix

Test matrix	
Tube geometry	
Outer diameter	12.78 mm
Inner diameter	10.98 mm
Thickness	1.8 mm
Heated length	240 mm
Vertical upward flow	
Pressure	101.3 kPa (1 atm)
Mass flux	100 ~ 500 $\text{kg/m}^2\text{s}$
Inlet temperature	50 °C

In this study, three concentrations (0.2, 0.4, 0.6%) of TSP and four concentrations (0.2, 0.4, 0.6, 0.8%) of boric acid were used.

2.2. Experimental procedure

The test loop was filled with plain water, TSP solutions, and boric acid solutions. After that, removing non-condensable gas is needed first. While the experiment is performed, the inlet temperature and the mass flow are maintained at desirable level.

Heating power in the test section was gradually increased by slowly increasing the voltage. The heat flux increment is about 20 kW/m^2 near the expected CHF onset point and 50 kW/m^2 in other section. Every time the voltage is increased, it is needed to check the stability of fluid condition of the test loop. It is the CHF onset point that the sudden increase ($\sim 70^\circ\text{C}$) of wall temperature of the test section happens. When the onset of CHF is observed, switch of the power supply have to be turn off to prevent the test section from breaking

immediately. The experiments were performed at inlet subcooling temperature of 50°C and five mass flux levels (100, 200, 300, 400, and 500 kg/m²s).

2.3. Results and Discussion

In case of TSP, 21.4% enhancement of CHF was observed at inlet subcooling temperature of 50 °C and extremely low mass flux (100 kg/m²s). The trend that the CHF enhancement ratio decreases with the increase of mass flux was shown, and except two cases (200, 300 kg/m²s), the CHF enhancement ratio decreased with the increasing of concentration. In case of boric acid, 14% enhancement of CHF was observed at inlet subcooling temperature of 50 °C and extremely low mass flux (100 kg/m²s).

The main effects of TSP are the decrease of surface tension and the increase of wettability. It is shown in the study of Jeong et al. [3] well. In case of boric acid, it is similar with TSP. From previous contact angle measurement, the effect of boric acid was demonstrated. It is confirmed the contact angle decrease with the increase of concentration of boric acid, that is, boric acid also has the effect which reduces surface tension. But it can be inferred that the effect on the CHF of boric acid is smaller than that of TSP because the decrease rate of boric acid is smaller than that of TSP.

These additives are not affect other properties except surface tension in low concentration (<1%), and the increase of wettability and shorter wave length on the interface between liquid and vapor are due to the decrease of surface tension. It is followed by the phenomena like the decrease in bubble diameter, breakup of bubbles and avoidance of bubble coalescence, and the CHF is enhanced. In Figure 1&2, the enhancement of CHF is shown well.

The effect of surfactant on the CHF depends on the competition of the possibility of instability occurrence and the increase of wettability which are due to the decrease of surface tension. At flow rate of 100~200 kg/m²s, flow regime is annular flow. In this case, the CHF mechanism is liquid film dryout (LFD). It is well known that the dryout in annular flow relates with droplet entrainment and deposition. In this region, the mechanism of the CHF enhancement is not clear yet.

At flow rate of 300~500 kg/m²s, flow regime is slug flow. The CHF mechanism is departure from nucleate boiling (DNB). In this case, the CHF is occurred by instability of large slug. If the instability is once occurred around the unstable and wavy slugs, and the surface is dried out having no leeway rewetted. Then, the CHF is occurred. Because the reduced surface tension increases the instability the CHF can decrease rather than increase in this region.

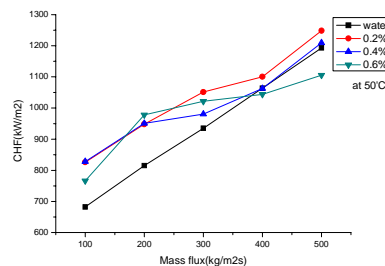


Figure 1. CHF as a function of mass flux for different concentrations of TSP

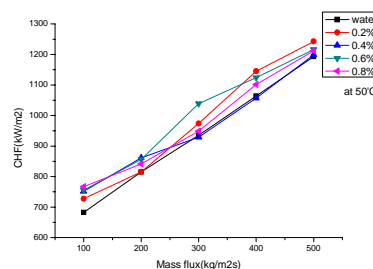


Figure 2. CHF as a function of mass flux for different concentrations of boric acid

3. Conclusions

In case of TSP, 21.4% enhancement of CHF was observed at inlet subcooling temperature of 50°C and extremely low mass flux (100 kg/m²s). The trend that the CHF enhancement ratio decreases with the increase of mass flux was shown, and except two cases (200, 300 kg/m²s), the CHF enhancement ratio decreased with the increase of concentration. In case of boric acid, 14% enhancement of CHF was observed at inlet subcooling temperature of 50°C and extremely low mass flux (100 kg/m²s). The CHF increasing rate decreased with the increase of mass flux.

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