# Development of a Lab-Scale Hybrid Acid Cooler for in Very High-Temperature Process-Heat Conditions

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# 1. Introduction

The Very High Temperature Reactor (VHTR) heat is transferred to a thermo-chemical hydrogen production process through an intermediate loop. Since nuclear hydrogen coupled components of H<sub>2</sub>SO<sub>4</sub> decomposition process will operate at a at high pressure (~2.0MPa) and high temperature (~900°C) and a large mechanical stress with a corrosive H<sub>2</sub>SO<sub>4</sub> and SO<sub>3</sub> environment, the development of nuclear hydrogen coupled components is one of the current unsolved issues in nuclear hydrogen generation[1]. Most of the sulfuric acid loops have an operational difficulty at a high pressure and high temperature condition because of the harsh corrosion characteristic of the high temperature sulfuric acid. Fragile quartz is an excellent component material at low pressure sulfuric acid loop but quartz is not allowed to use beyond the pressure of 0.3MPa at high temperature. KAERI (Korea Atomic Energy Research Institute) designed and constructed a small scale sulfuric acid experimental system which simulates a H<sub>2</sub>SO<sub>4</sub> decomposition part of the hydrogen production module of a Nuclear Hydrogen Development and Demonstration (NHDD) reactor [2]. The main usage of the loop is to screen and verify the candidates for the nuclear hydrogen coupled components. This experimental system needs a high-temperature sulfuric acid cooling system that is robust for a harsh environment owing to the corrosive chemicals with a very high temperature (>900°C).

### 2. Sulfuric acid loop

A sulfuric acid  $(H_2SO_4)$  loop is an open loop and consists of a  $H_2SO_4$  storage tank, a  $H_2SO_4$  feed pump, a pre-heater (evaporator), a  $H_2SO_4$  superheater (or  $H_2SO_4$  decomposer), a PHE(or SO<sub>3</sub> decomposer), a high-temperature cooler, a separator, a SO<sub>2</sub> trap, a low-temperature cooler, and a  $H_2SO_4$  collector (Figure 1).

Cold 96wt%  $H_2SO_4$  is superheated up to 500°C. In a superheating process,  $H_2SO_4$  steam starts decomposition into  $H_2O$  and  $SO_3$ , if it reaches the temperature of 400 °C. In the PHE, some fraction of the  $SO_3$  is decomposed into  $SO_3$  and  $O_2$ . In the high-temperature cooler,  $SO_3$  and  $O_2$  are formed into  $H_2SO_4$  gas and  $H_2SO_4$  gas and water vapor are condensed into a weaker liquid  $H_2SO_4$  than that at inlet of the loop. The toxic liquid  $H_2SO_4$  is separated in the separator and the sulfur dioxide in the mixture gas is removed in the NaOH trapping system. The design condition of the Sulfuric acid loop is as follows;

0	Working Fluid	Sulfuric acid
0	Design Temperature	950 °C
0	Design Pressure	1.0 MPa
0	Design Flow	1.0 kg/min

#### 3. Design and validation

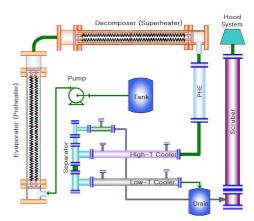


Figure 1 Schematics of a sulfuric acid loop

#### 3.1 Hybrid acid cooler design

Harsh corrosion characteristic of sulfuric acid on metal limits the design of a high-pressure acid cooler. Ceramic or Quarts has excellent corrosion resistance, But they have many manufacturing problems. Quartz is not allowed to use beyond the pressure of 3.0bar due to its fragile characteristics. In the case of ceramics, it allowed at 5~20 bar. But these kind of brittle materials have many difficulties to connect or join between components because they are easy to break especially when subjected to thermal stress.

As described in the previous section, when high temperature  $H_2O$  and toxic  $SO_3$  cool down, they react to each other and become sulfuric acid in the high-temperature cooler. We designed a hybrid acid cooler with two cooling zones; a air cooling zone for the hot side and a water cooling zone for the cold side (Figure 2). The sulfuric acid flow path composed of three parts in the cooler; a Teflon-lined outlet plenum, the SiC tubes and a Hastelloy C-276 inlet plenum which has good corrosion resistance in  $SO_3$  environments.

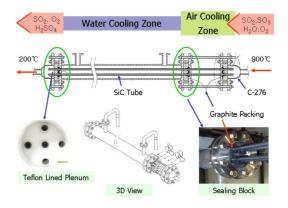


Figure 2. Hybrid acid cooler (Tin = 900°C, Tout = 200°C)

The design solutions are listed in Table 1 to overcome many design problems such as thermal stress, pressure sealing and metal corrosion. Because of the lack of thermodynamic and transport properties in the high-pressure and high-temperature region of sulfuric acid or mixture gas, clumsy calculation for sizing is inevitable and it lead to an excessive design margin of the cooler. The sizing of a 20kW acid cooler is provided by the Aspen plus chemical process simulator [3]. The design input is assumed that the pressure of 5bar, temperature of 150°C with 0.5 kg/min mass flowrate.

Table 1. Problems and solutions for design of an acid cooler

Problem	Key parts	Solution
Corrosion of	Materials	- Cold side: Teflon lined
the Sulfuric	Teflon,	stainless steel Plenum
acid (or	Ceramic,	- Cooling elements: SiC tubes
mixture gas)	C-276	- Hot side: C-276 Plenum + Ion
flow path		Beam Mixing (periodic
		maintenances are required)
Thermal	SiC tube	- Soft connection of the SiC
Stress of the		tubes by using a Graphite
cooler		compression packing
		- Hot side: air cooling
		- Cold side: water cooling
Pressure	SiC tube	- Multiple Graphite
Sealing	Sealing	compression packing in a C-
_	block	276 sealing block
Measurement	Thermo-	- Gold coating on the surface
	couples	- Periodic maintenances

#### 3.2 Validation test

We manufactured a Lab-scale (20kW) hybrid acid cooler and performed a pressure test up to 15bar (Figure 3). The overall performance test of the hybrid acid cooler is carried out in conjunction with two kinds of preliminary loop tests; a water (steam) test and a sulfuric acid test at atmosphere, respectively. The operating results of sulfuric acid loop are shown in Figure 4. Unfortunately the inlet temperature of the high-T cooler is very low for both tests. There is a need to raise the inlet temperature for a thermal performance test of the cooler. Teflon lined plenum and SiC tubes show very excellent results for corrosion. But the Hastelloy C-276 plenum, without coated by Ion-beam mixing, is observed a thin corrosion layer (Figure 3(b)

# 4. Conclusion

We designed and assembled a corrosion proof high-pressure and high-temperature hybrid acid cooler (Figure 4). We can draw the following conclusions for designing the cooler;

1. A control of the metal corrosion is a primary design concern for an acid cooler

2. Quartz material has a good corrosion resistivity but it is not suitable for a high pressure acid cooler application

3. Combination of a Teflon-lined plenum, the SiC tubes and a Hastelloy C276 plenum enables good corrosion resistance of a sulfuric acid cooler

4. We achieved a 15bar pressure test and accomplished both a high-temperature steam test and a sulfuric acid test for the present hybrid cooler

In the future, we will perform the validation tests of the hybrid cooler at higher temperature (> 100 C) condition and refine of the analytic design (thermal sizing) method for the high temperature sulfuric acid cooler with *Exothermic Reaction of*  $SO_3$  and  $H_2O$  and Condensation of  $H_2SO_4$  and  $H_2O$  vapor.

# ACKNOWLEDGEMENTS

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# REFERENCES

[1] J. H. Chang et al., A study of a Nuclear Hydrogen Production Demonstration Plant, Nuclear Engineering and Technology, Vol. 39, No. 2, pp. 111-122, 2007.

[2] S. D. Hong et al., A High Pressure and High Temperature Sulfuric Acid Experimental System, Proc. KNS Autumn Meeting, 2008.

[3] Aspen Plus version 2006, Aspen Technology, 2007



(a) hybrid acid cooler

(b) C-276 plenum

Figure 3. Manufactured hybrid acid cooler and Hastelloy C-276 inlet plenum after 12 hours sulfuric acid test

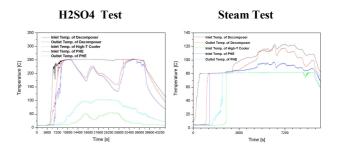


Figure 4. Operating results of a sulfuric acid test and a steam test