

## Neutron shielding characteristics of nano-B<sub>2</sub>O<sub>3</sub> dispersed Poly Vinyl Alcohol

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### 1. Introduction

Neutron is sometimes beneficiary to human beings while they are unwanted for most cases same as the other radiations such as gamma, beta, and alpha, etc. do. Shielding for neutrons therefore is extremely important to keep the radiation environment safe. Especially, it is critical to absorb (or shield) neutrons generated from the spent fuel in a container/storage, nuclear reactor, and cyclotron, etc. In this regard, light materials containing neutron absorbers such as borated-polymers are very useful to shield neutrons in those radiation environments.

This investigation is focused on the development of borated polymer-based materials whose neutron shielding efficiency is greatly enhanced by using nano-sized boron compounds. Boron is well known as a thermal neutron absorber due to its large thermal neutron absorption cross-section ( $\sigma_{th} = 760 \text{ b}$ ,  $b = 10^{-24} \text{ cm}^2$ ). Although absorption of neutrons in the medium is mainly dependent on the boron atomic weight concentration, we firstly observed the size of boron particles also has an important role in neutron shielding. Mean free path of neutrons colliding with the smaller particles dispersed in the medium might be decreased when it is compared to the larger particles at the same atomic weight concentration [1-3]. This means that the neutron shielding efficiency of a polymer mixed with the smaller boron compounds is higher than that of a polymer mixed with the larger boron compounds at the same atomic weight boron concentration.

### 2. Methods and Results

To verify the enhancement of neutron shielding efficiency dependent on the size of the boron compound, micro- and nano-boron compound containing composites were prepared and material properties were analyzed. Neutron shielding tests were also performed.

#### 2.1 Materials Preparation

As a neutron absorbing boron compound, B<sub>2</sub>O<sub>3</sub> (High Purity Chemicals, Inc., Japan) was used while PVA (Poly Vinyl Alcohol, Aldrich) was used as a polymer base. Nano-B<sub>2</sub>O<sub>3</sub> powder was produced by using a MA (Mechanical Activation) process. Initial micro-B<sub>2</sub>O<sub>3</sub> powder was grounded by a high energy ball mill under the various experimental conditions including metal ball to powder ratio, jar rpm and duration of jar revolution,

etc. Micro-sized (untreated) and nano-sized (MA treated) B<sub>2</sub>O<sub>3</sub> powders were mixed homogeneously with PVA powder under the given atomic boron concentrations. Mixed powders were then hot pressed at the preset pressures making various thicknesses of the borated-polymer samples. Table 1 shows the physical and geometrical characteristics of each sample.

Table 1. Characteristics of micro- and nano-B<sub>2</sub>O<sub>3</sub> dispersed PVA samples.

Properties	Micro-B <sub>2</sub> O <sub>3</sub>	Nano-B <sub>2</sub> O <sub>3</sub>	PVA
Size Distribution	200 $\mu\text{m}$ ~ 300 $\mu\text{m}$	0.1 $\mu\text{m}$ ~ 1 $\mu\text{m}$	~ 200 $\mu\text{m}$
Atomic Weight Concentration of Boron	1.0% & 2.5% in PVA		
Sample Thickness	0.2cm, 0.5cm, 0.75cm, 1.0cm		

#### 2.2 SEM and TEM Analyses

Degree of dispersion of the boron compounds and the average particle sizes in the prepared samples were observed by the various analytic means including SEM (Scanning Electron Microscope) and TEM (Transmitted Electron Microscope), etc. Images in Fig. 1 and Fig. 2 show that the distribution of the boron particles is homogeneous in the medium, and also the average sizes of the particles are (a) 200 ~ 300 $\mu\text{m}$  for a untreated sample and (b) 0.1 $\mu\text{m}$  ~ 1 $\mu\text{m}$  for a MA treated sample.

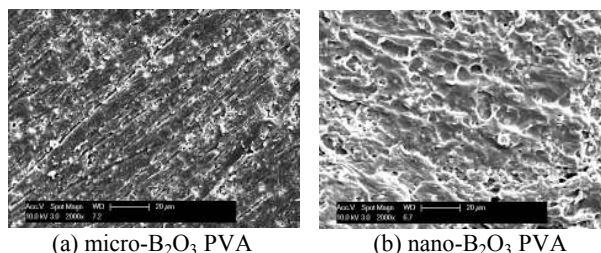


Fig. 1 SEM images showing distribution of the particles in the PVA medium (a) micro-B<sub>2</sub>O<sub>3</sub> dispersed PVA. (b) nano-B<sub>2</sub>O<sub>3</sub> dispersed PVA

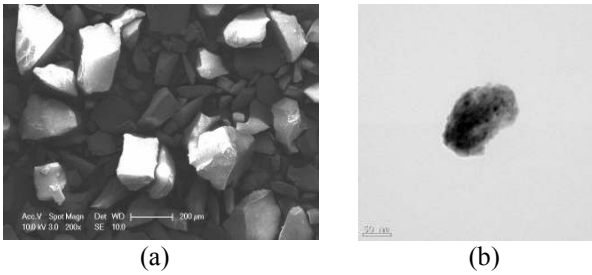


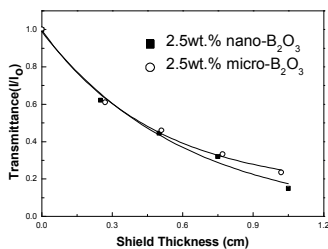
Fig.2. TEM images of the borated polymer with (a) micro-B<sub>2</sub>O<sub>3</sub> dispersed PVA and (b) nano-B<sub>2</sub>O<sub>3</sub> dispersed PVA.

### 2.3 Neutron Shielding Tests

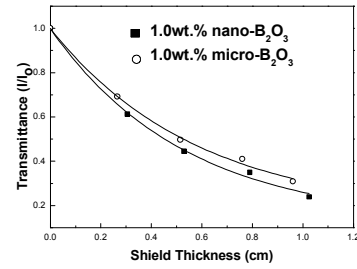
Neutron shielding efficiencies of each borated-PVA samples were tested by measuring the ratios of the incident neutron flux to the shield transmitted neutron flux. Samples with the various thicknesses were irradiated by the thermal neutrons guided from the FCD (Four Circle Diffractometer) beam port installed at HANARO reactor. Neutron wavelength and incident flux were  $\sim 0.997\text{\AA}$  and  $\sim 6.6 \times 10^5 \text{ n/cm}^2/\text{sec}$  at the sample position, respectively. A H-3 gases proportional counter whose effective detection length is 8.8cm was situated at  $\sim 2\text{m}$  away from the sample. Samples were installed at the outlet of the beam port. The duration of each measurement was 10sec and 10 measurements for each sample were averaged.

### 2.4 Particle Size Effects on Neutron Shielding

Thermal neutrons are absorbed (or moderated) by the relationship  $I(t) = I_0 e^{-\Sigma_{th} t}$ , where  $I_0$  is the incident neutron beam flux,  $t$  is the thickness of the medium, and  $\Sigma_{th}$  is the macroscopic thermal neutron absorption cross-section expressed by  $\Sigma_{th}(\text{cm}^{-1}) = n\sigma_{th}$ , where  $n$  is the number density and  $\sigma_{th}$  is the microscopic thermal neutron absorption cross-section of atomic boron. Usually  $\Sigma_{th}$  can be determined by the experimental measurements. Fig. 3 shows the thermal neutron transmittance in (a) 2.5wt% and (b) 1.0wt% boron dispersed PVA medium. Macroscopic thermal neutron absorption cross-sections are increased about 15% for nano-B<sub>2</sub>O<sub>3</sub> PVA compared to the micro-B<sub>2</sub>O<sub>3</sub> PVA. Macroscopic thermal neutron absorption cross-sections obtained from Fig. 3 are shown in Table 2.



(a)



(b)

Fig. 3. Neutron shielding efficiencies for micro-B<sub>2</sub>O<sub>3</sub> and nano-B<sub>2</sub>O<sub>3</sub> dispersed PVA with (a) 2.5 boron wt% (b) 1.0 boron wt%.

Table 2. Macroscopic thermal neutron absorption cross-sections ( $\Sigma_{th}$ ) measured for micro- and nano-B<sub>2</sub>O<sub>3</sub> dispersed PVA.

Atomic boron weight concentration	$\Sigma_{th}(\text{cm}^{-1})$ , micro-B <sub>2</sub> O <sub>3</sub>	$\Sigma_{th}(\text{cm}^{-1})$ , nano-B <sub>2</sub> O <sub>3</sub>
2.5 wt%	1.49	1.72
1.0 wt%	1.25	1.42

### 3. Discussions

In this investigation, we observed the enhanced neutron shielding efficiency when the smaller sized boron compounds were dispersed in the polymer matrix. Mean free path ( $l$ ) of the particles,  $l \sim (n\sigma_s)^{-1}$ , where  $n$  is the number density,  $\sigma_s = \pi d^2$  is the scattering cross-section, and  $d$  is the diameter of the particle, is assumed to be decreased by the factor of  $10^3$  when the diameter of the particle is  $10^3$  times decreased. Further research nevertheless is necessary to verify this phenomenon by applying to the other materials and also using a theoretical evaluation such as MCNP simulation in depth. In general, mechanical and thermal properties as well as electrical and optical properties are enhanced by using nano-particles containing polymer [4]. Consequently, it is promising to develop the efficient neutron shielding materials with less boron content or thinner shields maintaining neutron shielding capacity while the material properties are greatly enhanced.

### REFERENCES

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