The Optimizing Study for Separation Process of ⁶⁴Cu from ⁶⁷Ga Waste Product

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1. Introduction

The radioisotope of ⁶⁴Cu is suitable for applications [1] in both PET imaging and targeted radiotherapy owing to its unique decay scheme, which combines EC (41 %), β^{-} -decay (49 %, E max: 0.57 MeV), and β^{+} decay (19 %, E max: 0.66 MeV). A long enough halflife of 12.7 h with its annihilation radiation (0.511 MeV) makes it to study the kinetics of small and large molecular carriers with PET imaging and it is also considered as therapeutic potential due to Auger electron emission. The production of ⁶⁴Cu is readily achieved by proton bombardment on Ni coated on the copper plate target or chemical separation from ⁶⁷Ga waste products. In this study, it was found out that a reasonable quantity of ⁶⁴Cu is co-produced during the production of ⁶⁷Ga from enriched ⁶⁸Zn including many other radioisotopes (see table 1). Using anion exchange method, the chemical separation of ⁶⁴Cu from the waste products of 67 Ga is carried and report about hundred milicuries of 64 Cu could be achievable within 1 hour after EOB.

Table 1. Characteristic gamma-ray emissions for co-produced radioisotopes

Isotope	Emission (abundance, %)
⁶⁷ Ga	93.3(38.0), 184.6(24.0), 300.0(16.0)
⁶⁷ Cu	91.3(7.0), 93.4(16.0), 184.5(40.0)
⁵⁷ Ni	127.3(15.0), 511, 1377.6(84.9)
⁵⁷ Co	122.1(85.2), 136.5(11.1)
⁵⁵ Co	477.2(16.0), 931.0(73.0), 1048.3(18.0)
⁶⁴ Cu	511, 1345.8(0.48)

2. Methods and Results

2.1. Co-products of several radioisotopes during ⁶⁷Ga production

The 68 Zn target material was plated onto a nickel coated copper target [2]. The 67 Ga was produced via the 68 Zn(p,2n) 67 Ga reaction using a 200 uA beam of 30 MeV protons for 9 hours. After irradiation, the 67 Ga was separated from the 68 Zn target. The washed solution (called after "waste") 350 ml of the eluent contains a majority of the radioactive by-products (shown in Table 1), which was used in the present study. The solution was heated to evaporate and dissolved in 0.2M HCl in 96 % methanol and then transferred onto an anion exchange column. A gamma-spectrometer with HP Ge(Li) detector coupled to PC-

base MCA was taken from the solution as shown in fig.1, corresponding to characterized their emission and abundance for each product given in table 1.



Figure 1. A spectrum obtained from the waste of products after dryness. Gamma spectrometry of the ⁶⁷Ga waste showed that ⁵⁵Co, ⁵⁷Ni, and ⁵⁷Co, as well as ^{67/64}Cu are co-produced during the ⁶⁷Ga production. An established method was used both in the separation of these radioisotopes from the ⁶⁷Ga production and the purification of the ⁶⁸Zn target material [2].

2.2. Separation of ${}^{57}Ni$, ${}^{55/57}Co$ and ${}^{64}Cu$ from the waste solution

The separation method of metals with aqueous HCl solutions and anion exchange resins is well established [3]. For the separation sequence of ⁶⁴Cu from other radionuclides, similar procedures are employed– the metals forming anionic complexes are adsorbed on the anion exchange resin at high HCl concentration and eluted progressively with acid solutions.



waste solution.

The schematic steps for the chemical separation of 64 Cu from the waste solution are shown in fig.2. First, the separation of 57 Ni and $^{55/57}$ Co was achieved by successive washed of 0.2 M HCl in 96 % methanol (30 ml) and 1.3 M HCl in 55 % isopropyl alcohol (30 ml), respectively. Then the $^{64/67}$ Cu was eluted using 0.5 M HCl in 50 % isopropyl alcohol (30 ml), typical elution profile is shown in fig. 3



3. Results and discussion

The reasonable amount of ⁶⁴Cu (120 mCi) is expected to achievable with the present method at EOB. Figure 4 shows the gamma spectrum obtained from the final separated product, which indicates only copper nuclides of ⁶⁴Cu and ⁶⁷Cu and the rest of metals are removed completely. However, it is unable to separate from ⁶⁴Cu and ⁶⁷Cu. The production rate of ⁶⁴Cu is to be estimated about 500 mCi and the separation yield is estimated to 24 %, which has to be enhanced by optimized process.



Figure 4. The gamma spectrum obtained after separation. Copper radionuclides were separated from other metals.

Several factors are should considered to improve the quantity of products and to apply waste management with more convenient method. During the separation of the ⁶⁴Cu, the processing time required to evaporate the waste solution, which contained highly radioactive nuclides has to be reduced for protecting the operator and preventing the losses by natural decay. With these reasons, an optimized automatic system is proposed as indicated at Fig. 5.



Figure 5. The proposed automatic ⁶⁴Cu separation system

The variation in separation yields of ⁶⁴Cu radioisotopes extracted from the waste solution of ⁶⁷Ga is believed to be largely dependent on the HCl concentration and the loading cation exchange column during the process. The optimized condition has to be studied more for the next step.

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