

## Densification Behavior of BN-added UO<sub>2</sub>

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### 1. Introduction

Boron is commercially used as a neutron absorber fuel. A neutron absorber fuel is burned out or depleted during reactor operation. Westinghouse have been produced the Integral Fuel Burnable Absorber (IFBA) which is enriched UO<sub>2</sub> fuel pellets with a thin coating of zirconium diboride (ZrB<sub>2</sub>) on the outer surface. Standard sintered fuel pellets are sputter coated with ZrB<sub>2</sub>. It is known that IFBA fuel can incur 20% to 30% additional fabrication costs. [1]

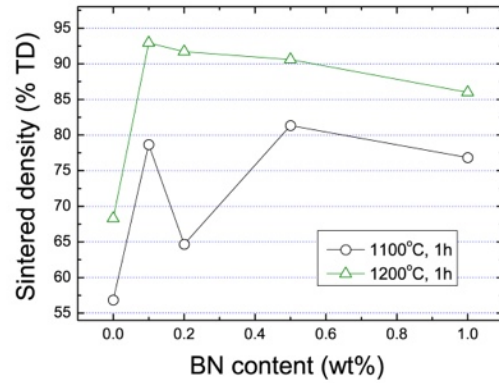
Boron-dispersed UO<sub>2</sub> fuel pellet made by the conventional pressing and sintering process of a powder mixture of UO<sub>2</sub> and B compound might be more cost-effective than IFBAs. M. G. Andrew et al. [2] tried to sinter boron-dispersed UO<sub>2</sub> green pellet. However, they reported that boron-dispersed UO<sub>2</sub> fuel pellet is very difficult to be fabricated with a sufficient level of boron retention and high sintered density (greater than 90 % of theoretical density) because of the volatilization of boron oxide.

We have investigated the densification behavior of mixtures of UO<sub>2</sub> and various boron compounds, such as B<sub>4</sub>C, BN, TiB<sub>2</sub>, ZrB<sub>2</sub>, SiB<sub>6</sub>, and HfB<sub>2</sub>. Boron compounds seemed to act as a sintering additive for UO<sub>2</sub> at a certain low temperature range. [3-5]

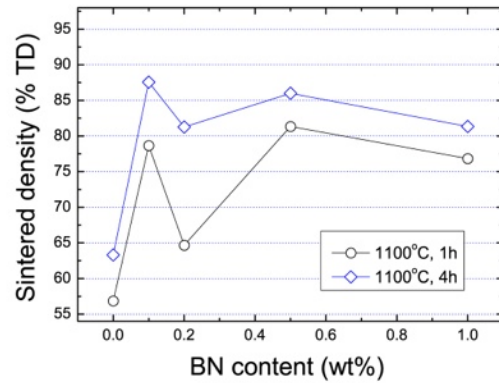
In this study, the densification behavior of BN-added UO<sub>2</sub> pellet has been investigated by sintering green pellets of a mixture of UO<sub>2</sub> powder and BN powder in H<sub>2</sub> atmosphere.

### 2. Experimental

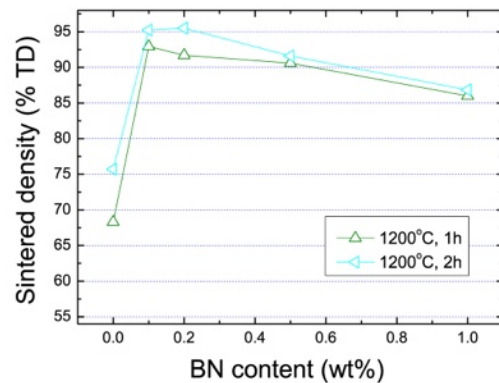
ADU route UO<sub>2</sub> powder and 70 nanometer-sized BN powder were used for sample preparation. UO<sub>2</sub> powder and BN powder were mixed and crushed in a ball mill for 24 h using an ethanol medium. A dried powder mixture was granulated with a 30 mesh sieve. The granules were mixed with a 0.3 wt% of zinc stearate in a tumbling mixer for 30 min. The compaction was conducted in a single acting press under about 3 ton/cm<sup>2</sup>. Green pellets were sintered at 1100 °C and 1400 °C for 1 to 4 h in a H<sub>2</sub> atmosphere. The sintered density was measured by a water immersion method. Microstructures were observed using an optical microscope after polishing the cross-section of the sintered pellet up to a 1 μm diamond polish. Residual boron content in the sintered pellets was measured by Prompt Gamma Activation Analysis (PGAA) at HANARO.



(a)



(b)



(c)

Fig. 1. Sintered densities of BN containing UO<sub>2</sub> pellets with various BN contents.

### **3. Results**

Measured sintered densities of sintered pellets with various initial BN contents were represented in Fig. 1 according to the sintering temperature and time. It appears that the densification of BN-added UO<sub>2</sub> pellet is significantly enhanced compared with that of pure UO<sub>2</sub> pellet. Even an addition of 0.1 wt% BN considerably increases the sintered density, and then the sintered seems to gradually decrease with increasing the BN addition. It might be attributed to the volatilization of excess boron oxide. BN-added UO<sub>2</sub> pellet can be densified up to about 95 % of theoretical density at 1200 °C for more than 1 h in a H<sub>2</sub> atmosphere.

### **4. Conclusions**

A high density BN-added UO<sub>2</sub> pellet can be fabricated after sintering at 1200 °C for more than 1 h in a H<sub>2</sub> atmosphere. The sintered density of BN-added UO<sub>2</sub> pellet can be increased up to about 95 %TD. Enhanced densification behavior of BN-added UO<sub>2</sub> pellet might be attributed to fast material transfer through the boron oxide liquid phase which is formed by the decomposition of BN. An appropriate amount of boron compound seemed to have an effect on the densification enhancement of UO<sub>2</sub> pellet during sintering at a certain low sintering temperature in a H<sub>2</sub> atmosphere.

### **ACKNOWLEDGMENTS**

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