# Preliminary Plugging tests in Narrow Sodium Channels by Sodium and Carbon Dioxide reaction

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#### 1. Introduction

Printed circuit heat exchanger (PCHE) is considered for the supercritical CO<sub>2</sub> Brayton cycle power conversion system of sodium-cooled fast reactors (SFRs), which is known to have potential for reducing the volume occupied by the sodium-to-CO<sub>2</sub> exchangers compared to traditional shell-and-tube heat exchangers [1, 2]. This report is on the investigation of the physical/chemical phenomena that a slow loss of CO<sub>2</sub> inventory into sodium after the sodium-CO<sub>2</sub> boundary failure in PCHEs in realistic operating conditions. The first phenomenon is potential channel plugging inside the narrow PCHE channel. Unlike a conventional shelland-tube type HXs, failures in a PCHE are expected to be small cracks. If the faulted channel is blocked, it may have a positive function for plant safety because the pressure boundary would automatically recover due to this self-plugging. The other one is damage propagation on pressure boundary, which is referred to as potential wastage with combined corrosion/erosion effect. Here, we report our preliminary experimental results of plugging tests.

#### 2. Methods and Results

## 2.1 Experimental Facility

Plugging experiments were carried out in the facility as reported previously (Figure 1) [3].

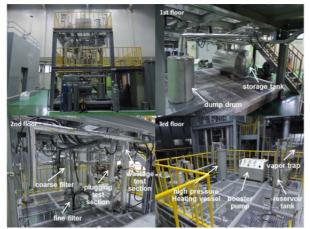


Fig. 1. Facility for the sodium- $CO_2$  interaction test (A booster pump and a high pressure heating vessel were added on 3rd floor recently).

## 2.2 Experimental Procedure

Plugging experiments for the reaction of sodium and  $CO_2$  were carried out in the facility as reported previously (Figures 2 and 3) [3].

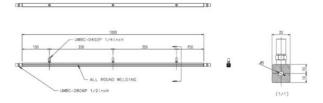




Fig. 2. Drawing of plugging test section (above) and its equipment in the facility (below [3]).

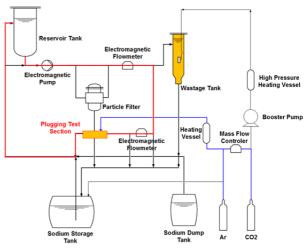


Fig. 3. Schematic diagram for the plugging experiment (red: sodium flow, blue: CO<sub>2</sub>).

# 2.3 Experimental Results

Flow rates of sodium with  $CO_2$  injection into the plugging test section (single sodium channel) in at 300–400 °C are shown Figure 4. Flow rates of sodium

in both 3 and 5 mm (inner diameter) of single sodium channel were slower as CO<sub>2</sub> injection at 300–400 °C. Flow rates of sodium in 3 mm channels at 300-400 °C decreased earlier and faster compared to 5 mm channels. In 3 mm of sodium channels, flow rates of sodium decreased immediately as CO<sub>2</sub> injection and reached at 0 m/s at 300 and 350 °C. In 5 mm of sodium channels, flow rates of sodium decreased as CO2 injection similar to those in 3 mm sodium channels. However, they reached at 0 m/s for 33, 350 and 75 min at 300, 350, and 400 °C, respectively. In both 3 and 5 mm of sodium channels, temperatures of sodium at inlet and outlet of the test section decreased when flow rates of sodium decreased at 300-400 °C. These results are not consistent with previous studies about reaction of sodium and CO<sub>2</sub>, in which temperature of sodium increased by exothermic reaction of sodium and CO<sub>2</sub> [4, 5]. Analysis of these results is of interest in this study.

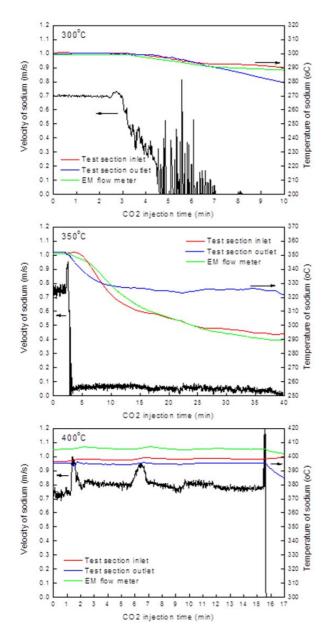


Figure 4. Flow rates of sodium in 3 mm channel with CO<sub>2</sub> injection into the plugging test section.

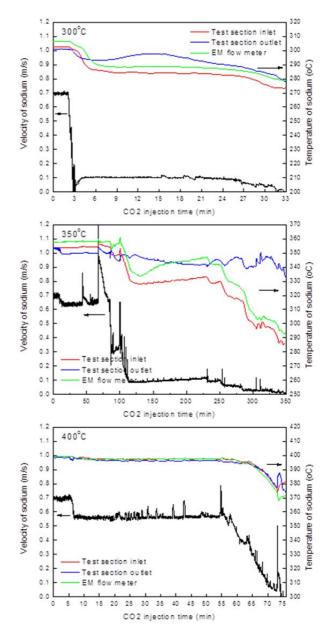


Figure 5. Flow rates of sodium in 5 mm channel with  $CO_2$  injection into the plugging test section.

It is known that reaction products such as sodium carbonate  $(Na_2CO_3)$  and amorphous carbon which are hardly soluble in sodium plug sodium channels. Proposed mechanism for reaction sodium and  $CO_2$  is as follows [6]:

$$\begin{split} &2 Na(s, l) + CO_2(g) \to Na_2O(s, l) + CO(g) \\ &Na_2O(s, l) + CO_2(g) \to Na_2CO_3(s, l) \\ &2 Na(s, l) + CO(g) \to Na_2O(s, l) + C(s) \\ &4 Na(s, l) + 3CO_2(g) \to 2Na_2CO_3(s, l) + C(s) \\ &4 Na(s, l) + CO_2(g) \to 2Na_2O(s, l) + C(s) \\ &2 Na(s, l) + 2CO_2(g) \to Na_2CO_3(s, l) + CO(g) \\ &C(s) + CO_2(g) \to 2CO(g) \end{split}$$

Above 400 °C, 100  $\mu$ m of nozzle completely shrank by thermal expansion of the test section (stainless steel 316). During sodium is heated to the temperature of the experimental condition, Ar gas was fed into the test section through the nozzle to prevent the nozzle from plugging. Ar gas was ordinarily supplied up to ~410 °C, whereas Ar gas hardly injected and then stopped when the temperature of the test section at ~410 °C. These results imply that micro cracks (< 100  $\mu$ m) at pressure boundary of sodium-CO<sub>2</sub> might be automatically clogged above 410 °C.

## 3. Conclusions

Physical/chemical phenomena that a slow loss of CO<sub>2</sub> inventory into sodium after the sodium-CO<sub>2</sub> boundary failure in printed circuit heat exchangers (PCHEs) were investigated. Our preliminary experimental results of plugging show that sodium flow immediately stopped as CO<sub>2</sub> was injected through the nozzle at 300–400 °C in 3 mm sodium channels, whereas sodium flow stopped about 60 min after CO<sub>2</sub> injection in 5 mm sodium channels. These results imply that if pressure boundary of sodium-CO<sub>2</sub> fails in a narrow sodium channel ( $\leq$  3 mm) would be plugged by reaction products in a short time whereas a relatively wider sodium channel ( $\geq$  5 mm) would be plugged with higher concentration of reaction products.

#### REFERENCES

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