# **Effect of resintering in Cr-doped UO<sup>2</sup> pellets**

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### **1. Introduction**

Many researchers are focusing on the development of large-grain  $UO<sub>2</sub>$  pellets as they can reduce corrosive fission gas release and thus enhance fuel reliability. It is known that an additive doping technology is effective on the grain growth of  $UO<sub>2</sub>$  pellets. Among them,  $Cr_2O_3$ -doped  $UO_2$  pellet is one of the promising candidates. To increase the grain size effectively, it is important to control the additive content and sintering atmosphere. Relevant research on the  $Cr_2O_3$  doped  $UO_2$ system revealed that the doped  $Cr_2O_3$  formed a liquid phase under optimized oxygen potential, and those liquid phases promoted the grain growth [1].

In-reactor densification of  $UO<sub>2</sub>$  became of major concern in 1972 when previously unknown fuel rod deformations had been observed in several reactor plants [2]. The in-reactor densification of  $UO<sub>2</sub>$  pellets has resulted in a shortening of the fuel column, the formation of gaps between pellets in the fuel column and between the pellets and the cladding. The damage imposed operating constraints on some reactors [3].

Evaluation of the in-reactor densification of the fuel pellets is important. The in-reactor densification of fuel pellets is commonly estimated using a resintering test, because In-pile tests are rather prolonged, complex, and expensive.

In this study, we investigated the effect of resintering in  $Cr_2O_3$  doped  $UO_2$  pellets. The test pellets obtained by sintering in a controlled wet atmosphere were resintered in dry hydrogen. The results showed that the grain size as well as pellet density were increased after the resintering. In particular, the grain size was greatly increased after the resintering, and the increase of grain size was discussed in terms of solubility of Cr in UO<sub>2</sub>.

### **2. Experimental**

The starting materials were  $UO<sub>2</sub>$  powder produced through the ADU (Ammonium Di-Uranate) process.  $Cr_2O_3$ -doped UO<sub>2</sub> was prepared by using  $Cr_2O_3$  powders and  $UO<sub>2</sub>$  powers. These powders were mixed for 12h using tumbling mixer. The contents of the  $Cr_2O_3$  were cintrolled to be 1200 and 1500ppm in weight.

The prepared  $Cr_2O_3$  containing  $UO_2$  powder mixtures were pressed into green pellets at  $3 \text{ ton/cm}^2$ . The green pellets were sintered at 1700°C for 6h in flowing mix gas of  $H_2$  and  $CO_2$ . In step-wise sintering process, the

oxygen potential of sintering gas was increased gradually to control the Cr dissolution rate.

A resintering test of sintered pellets was performed at 1700 $^{\circ}$ C for 24h in flowing H<sub>2</sub>. The density of sintered and resinterd  $UO<sub>2</sub>$  pellets was measured by a pycnometer. The pellets were sectioned axially, ground and polished. The polished pellets were thermally etched at  $1290^{\circ}$ C in carbon dioxide in order to examine their grain boundaries. The grain structures were examined by an optical microscope and the grain size was determined by the linear intercept method.

#### **3. Results**

Fig. 1 shows pore structure in  $Cr_2O_3$ -doped  $UO_2$ pellets  $(\mu g(Cr)/g(U)=1200)$  after (a) sintering and (b) resintering. Fig. 1(a) shows that most of the pore consists of small things, and partially large pores are distributed. The number of pores greatly decreased by resintering in dry  $H_2$  (Fig.1 b). The pore is transformed into lattice vacancies. The vacancies generated in this event migrate to the nearby pores, or to the grain boundaries. Vacancy migration to the grain boundary results in densification of the fuel.

However, in case of  $Cr_2O_3$ -doped  $UO_2$  pellets  $(\mu g(Cr)/g(U)=1500)$ , pore structure of sintered and resintered pellets is similar(Fig.1 c and d). It seems that because Cr inhibits the vacancy migration with increasing the Cr contents in  $UO<sub>2</sub>$  pellets, densification hardly occurred.



Fig. 1 Pore structure of  $Cr_2O_3$ -doped  $UO_2$  pellets after sintering and resintering test



Fig. 2 Grain structure of  $Cr_2O_3$ -doped UO<sub>2</sub> pellets after sintering and resintering test



Fig. 3 Comparison of grain size change in  $Cr_2O_3$  doped UO<sup>2</sup> pellets after a resintering test

Fig. 2 and 3 show the change of grain size before and after the resintering test in  $Cr_2O_3$  doped  $UO_2$  pellets. It shows that the grain size was greatly enlarged after resintering. In the case of 1200ppm of the  $Cr_2O_3$  doped  $UO<sub>2</sub>$ , the grain size was increased to about 5-times larger after resintering in dry hydrogen. Since the normal grain growth curve as a function of time shows a parabolic shape, this increase of grain size after resintering is abnormal.

It is known that the Cr solubility in  $UO<sub>2</sub>$  is decreased with a decrease in oxygen potential of the annealing gas. Since the sintering proceeded under wet hydrogen and resintering was under dry hydrogen, it is expected that a part of the dissolved Cr in the sintered  $UO<sub>2</sub>$  is precipitated or segregated during the resintering process. Therefore, it is considered that segregated or precipitated Cr plays an important role in the grain growth of pellets during the resintering process.

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