# The Effect of U<sub>3</sub>O<sub>8</sub> Powder on the Sintered Density of UO<sub>2</sub> Pellet I. Oxidation Temperature

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### 1. Introduction

 $UO_2$  is the most widely used nuclear fuel for current nuclear power generation. In addition, the dependence of nuclear power in the total power generation is growing due to eco-friendly factors, such as the regulation of  $CO_2$ emissions. Because of the limitations of uranium reserves and an increasing use of uranium resources, uranium price is increasing. The recycling of uranium resources is environmentally friendly as well as economical.

During the manufacturing process of  $UO_2$  pellets, the accompanying amount of scrap is approximately 8%. These scraps under an air atmosphere at constant temperature are recycled into  $U_3O_8$ . In general, the sintered density of  $UO_2$  pellet decreases and pore becomes coarse by the addition of  $U_3O_8$ . In other words,  $U_3O_8$  is a density controller as well as a pore-former

In this study, the influence of  $U_3O_8$  powder, formed by the various oxidation temperatures, on the sintered density of CANDU-type UO<sub>2</sub> pellet was investigated.

### 2. Methods and Results

CANDU-type pellet(97%TD, ~8um) into a platinum crucible was heated at 4  $^{\circ}$ C per minute in a muffle furnace and the oxidation time was constant at 10 hours. Oxidation was carried out in the 6 temperature conditions (350, 400, 450, 500, 600, 700  $^{\circ}$ C). XRD analysis showed that oxidized powder formed by the oxidation temperature was confirmed as U<sub>3</sub>O<sub>8</sub>. (Fig. 1)



Fig. 1. XRD pattern according to the oxidation temperature

5wt%  $U_3O_8$  powder which is formed by the temperature was added to ex-ADU natural UO<sub>2</sub> powders. Next those powders were mixed for 30 min. in a tubular mixer with 0.2wt% Acrawax for a lubricant. The pressure of compacts was carried out under 3 conditions (150, 300, 450 MPa). The green density was measured by the geometric method. Each density of compacting pressure (150MPa, 300MPa, 450MPa) was estimated as 4.6 ~ 4.7g/cm<sup>3</sup>, 5.3 ~ 5.4g/cm<sup>3</sup> and 5.8 ~ 5.9g/cm<sup>3</sup>, respectively. The green pellets were sintered at 1700 °C for 4 hours in a H<sub>2</sub> atmosphere. The sintered density of the pellet was measured by an immersion method.

Fig. 2 shows a change in sintered density by the addition of  $U_3O_8$  powder(5wt%) formed by each oxidation temperature.



Fig. 2. Sintered density according to the oxidation temperature

As shown in Fig. 2, the sintered density of  $UO_2$  pellet decreases as the oxidation temperature increases regardless of the compacting pressure. This is thought to be caused by the characteristics of  $U_3O_8$  powder that formed by the oxidation temperature. It was reported the higher oxidation temperatures increase the bulk density of the powder but the specific surface is decreased [1]. Namely, it is considered that higher oxidation temperatures cause a coarsening of  $U_3O_8$  particles.

#### 3. Conclusion

- XRD analysis showed that oxidized powder that was oxidized from 350  $^\circ$ C to 700  $^\circ$ C was confirmed as U<sub>3</sub>O<sub>8</sub>.

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- Increasing of the oxidation temperature of  $\rm U_3O_8$  powder causes decreasing of the sintered density of  $\rm UO_2$  pellet.

## REFERENCES

[1] S.H. Na et al., Proc. of the Korean Radioactive Waste Society, Autumn 2011, Jeju, 9(2) 2011, pp.163-164