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Introduction

- Last 20 years, many researches have been carried out to apply Ionic Liquids.
- While the need for organic solvents increases, ionic liquids have weakness on the high costs.
- These days DES (Deep Eutectic Solvent) which is similar to ILs is emphasizing and also introduced on Abbot et al (2003).
- Basically, DES is a mixture of two or more compounds, and the mixture has a lower melting point than each compound.
- DES's physical, chemical properties are similar with [6] existing ILs. However, DES has more sustainable advantages than ILs, not just synthesis method, low cost, non-toxic but also environmental and economic advantages [6,7,8].
- In this experiment, using DES based on glycerol, the experiment about dynamic characteristics as a heat transfer medium was conducted.

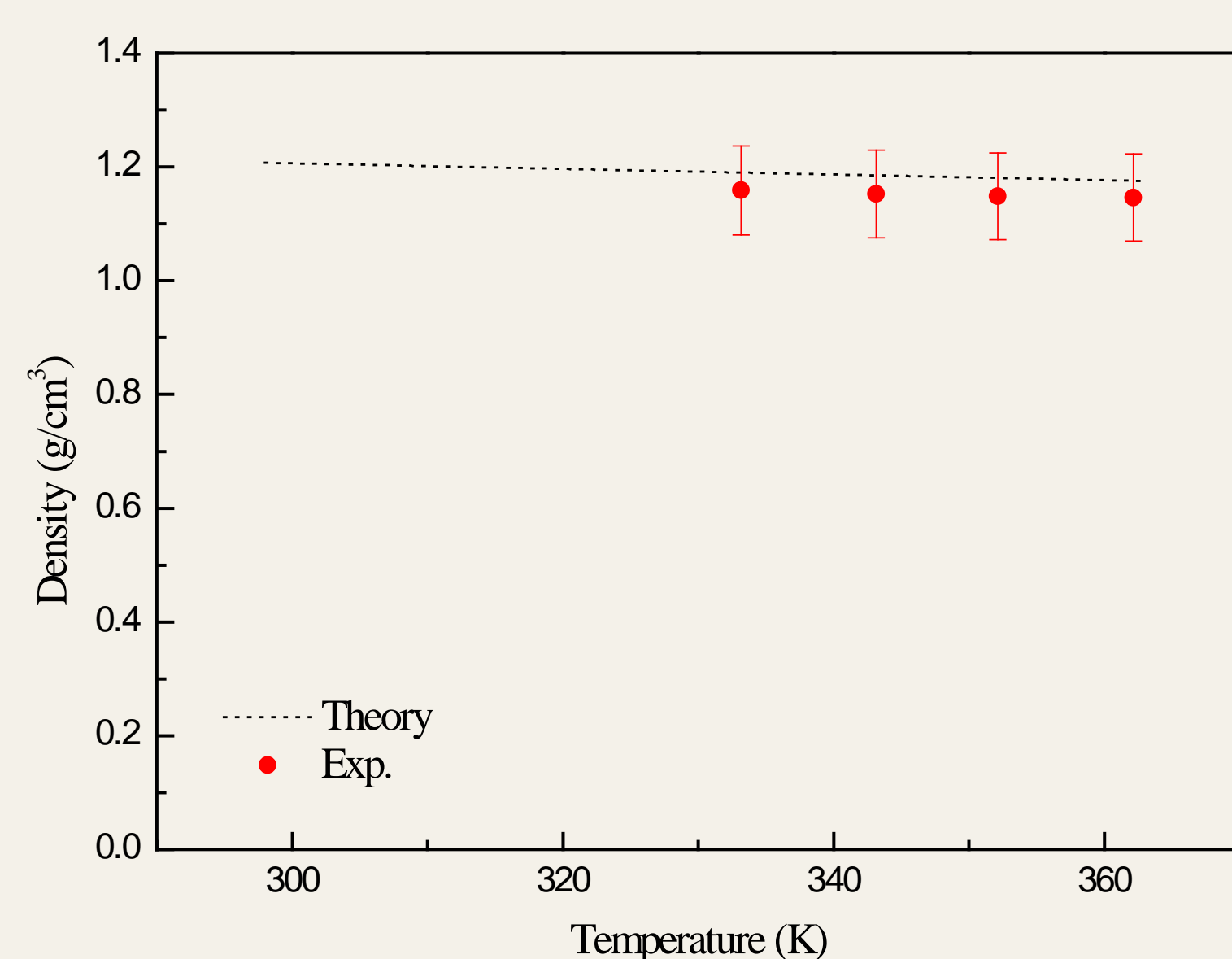
Preparations and Methods

Manufacturing DES

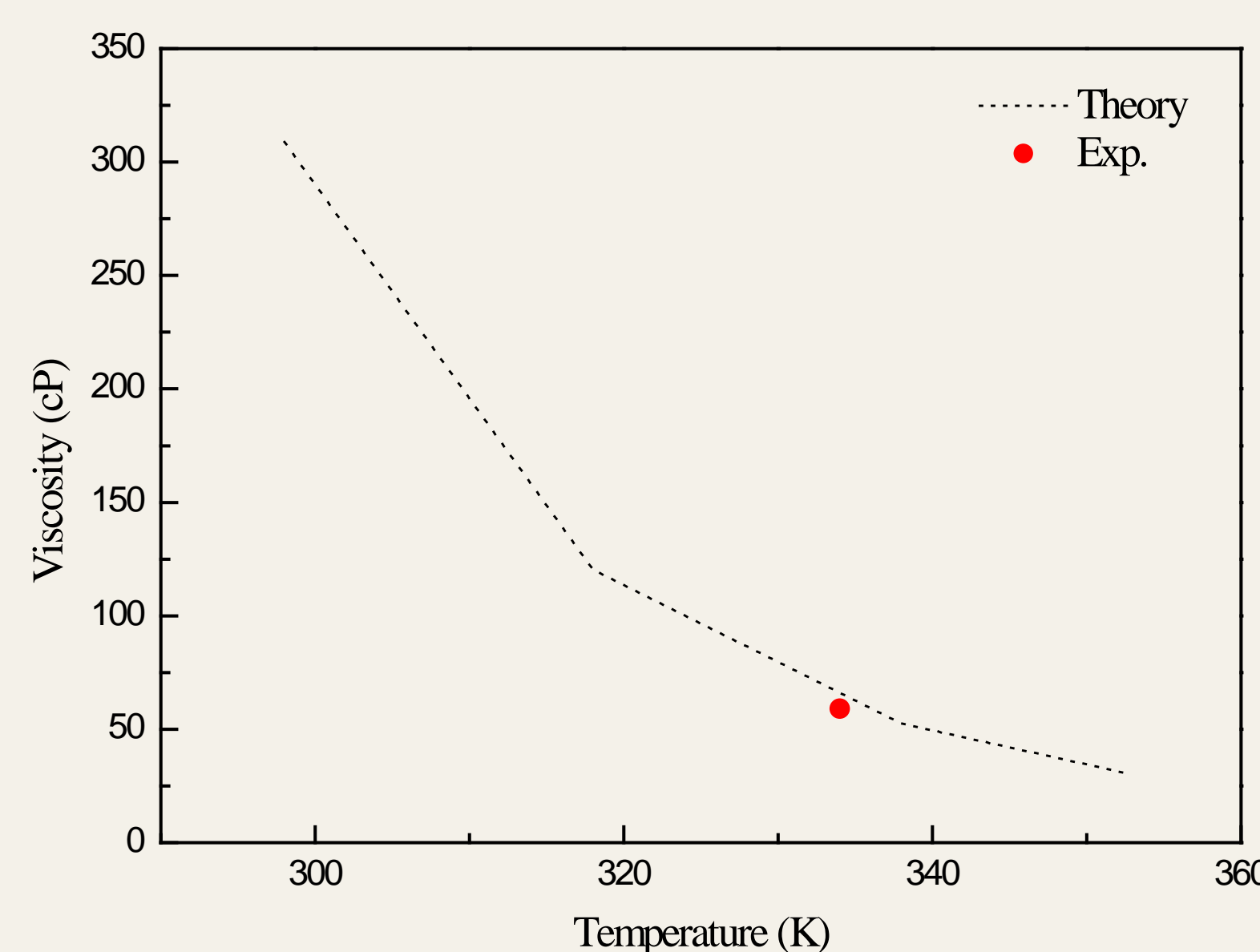
- Glyceline, which is DES, is a mixture of CHCl and Glycerol, 1:2 molar ratio will be mentioned.
- By using a 1:2 molar ratio of CHCl as a salt and Glycerol as a HBD, put them all together in the beaker or flask, then using the heated stirrer, DES is made by mixing at 75°C, 600 RPM for 2 hours.



- Chemical formula of CHCl is C₅H₁₄ClNO, melting point is 302°C, molecular mass is 139.62 g/mol. Chemical formula of Glycerol is C₃H₈O₂, melting point is 17.8°C, molecular mass is 92.094 g/mol. Following, manufacturing 2 kg of Glyceline, mass ratio of CHCl and Glycerol is 862.363 g : 1137.637 g.
- After measuring density and viscosity of manufactured Glyceline, then compared with Glyceline's properties which were known by M. K. Alomar et al (2016) [17].



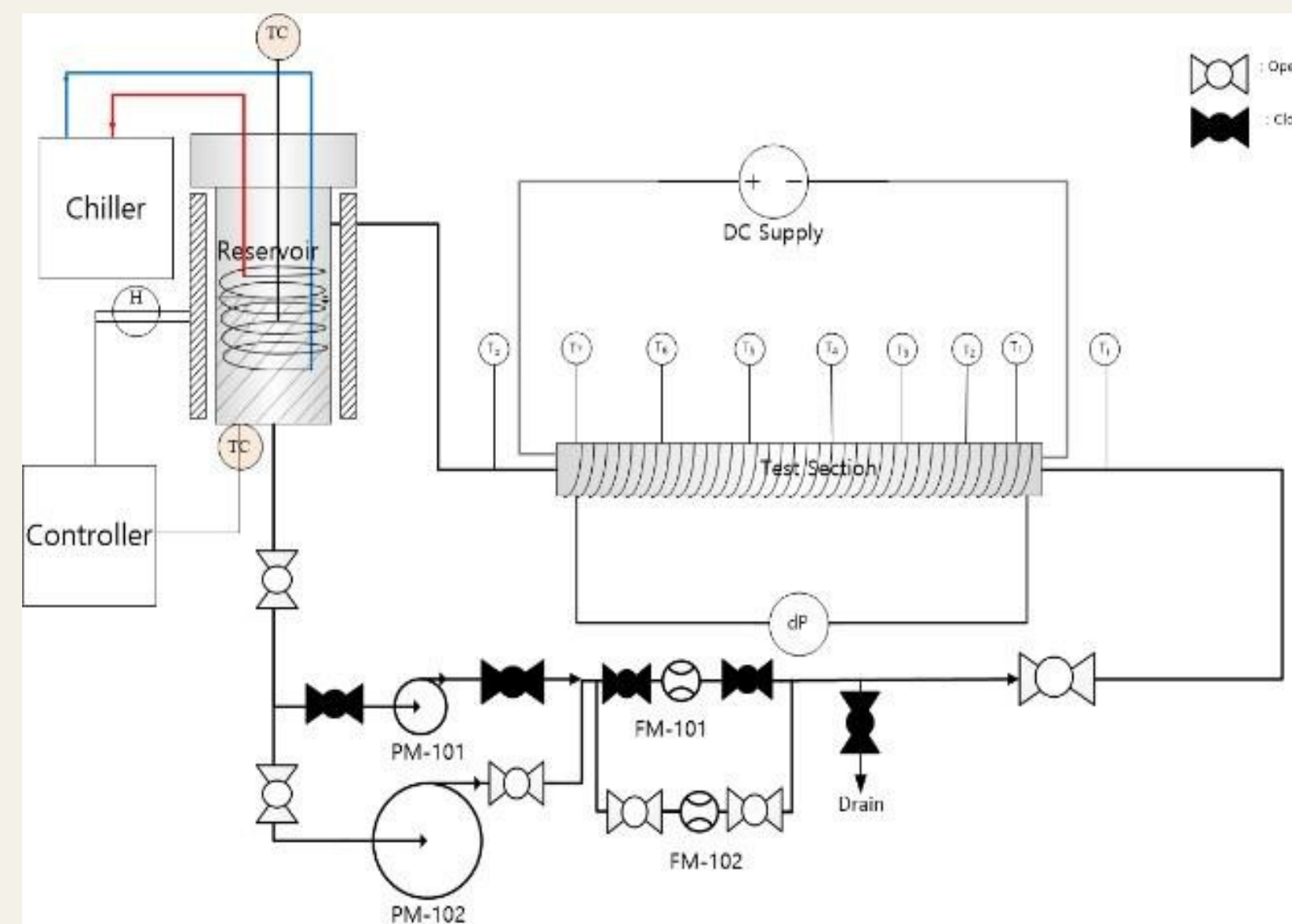
Preparations (cont.)



- Density of manufactured Glyceline for the main experiment is calculated as an average 2.5% lower than the theory value. If reflecting the uncertainty of measuring volume by using a reservoir and measuring mass by using scales, the theory value is in the measurement error range.
- Viscosity was measured directly by using a viscometer. It was measured at 334K, with an error between the theory value of 3.708%. The reason why there occurs a little error is during manufacturing DES, uncertainty of measuring mass causes uncertainty of molar ratio.
- Density and viscosity of manufactured Glyceline has a little gap between the theory value, through that this DES was manufactured appropriately to carry out this experiment.

Experimental apparatus

- To measure the Glyceline's dynamic characteristics, the experimental apparatus



- During dynamic characteristics experiment, calculate the DES's friction factor by using the data from the differential pressure gauge which is installed in the test section and the flux from the flow meter which is installed following the pump.

Result

- Put DES on the reservoir and circulate DES while keeping the steady temperature and same flux.
- After measuring the flux and differential pressure at this moment, calculate the friction factor according to the Reynolds number change, and compare with the Poiseuille friction factor and Blasius friction factor, which are theoretical values of the friction factor for water in a circular tube. The following equation shows how to calculate the Reynolds number by using the following equation.

$$Re = \frac{\rho v d}{\mu}$$

- The friction factor can be calculated by using the measured differential pressure and Darcy's formula. Darcy's formula is as follows.

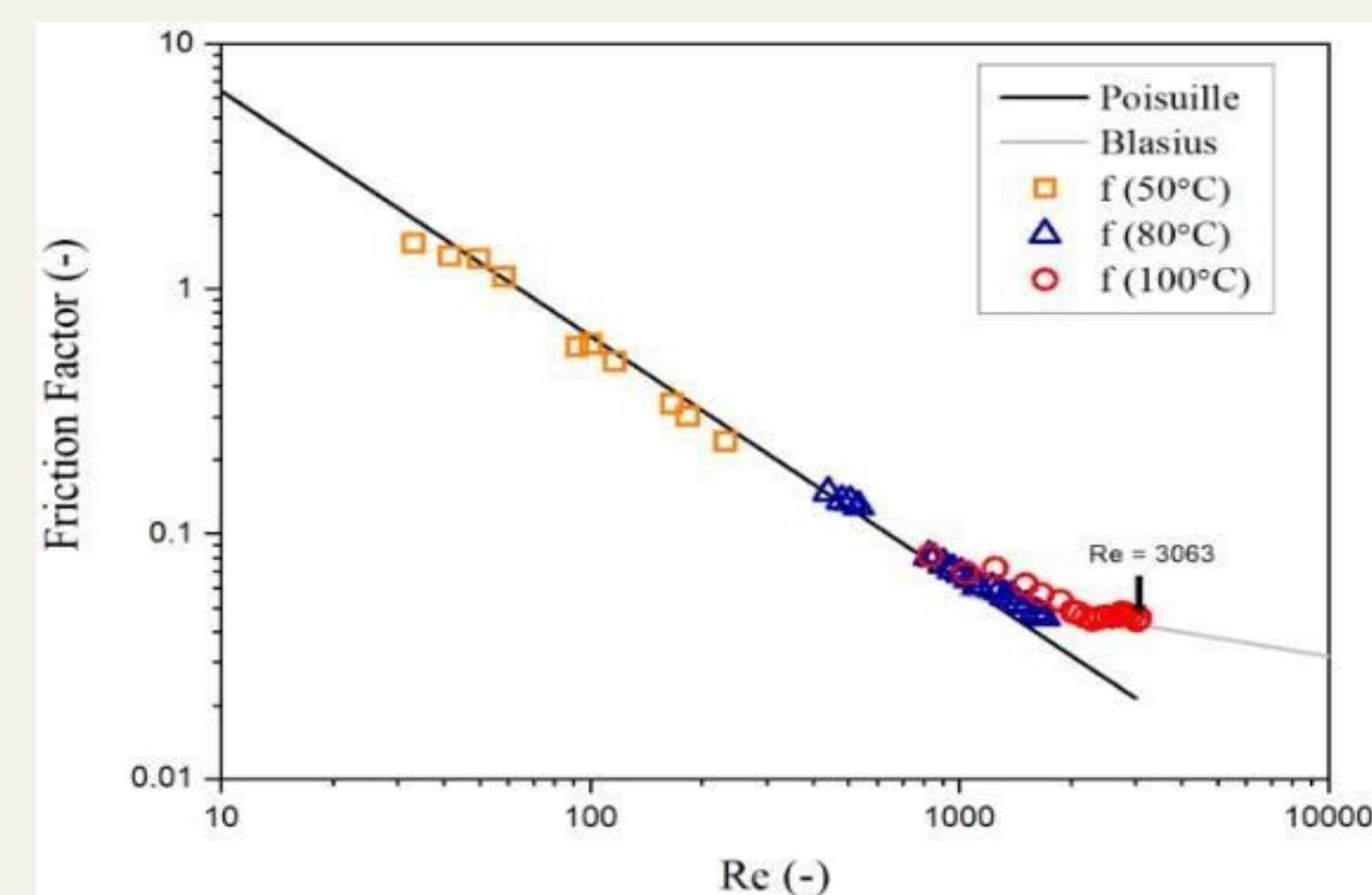
Result (cont.)

$$\Delta P = f \frac{L}{D} \frac{1}{2} \rho v^2$$

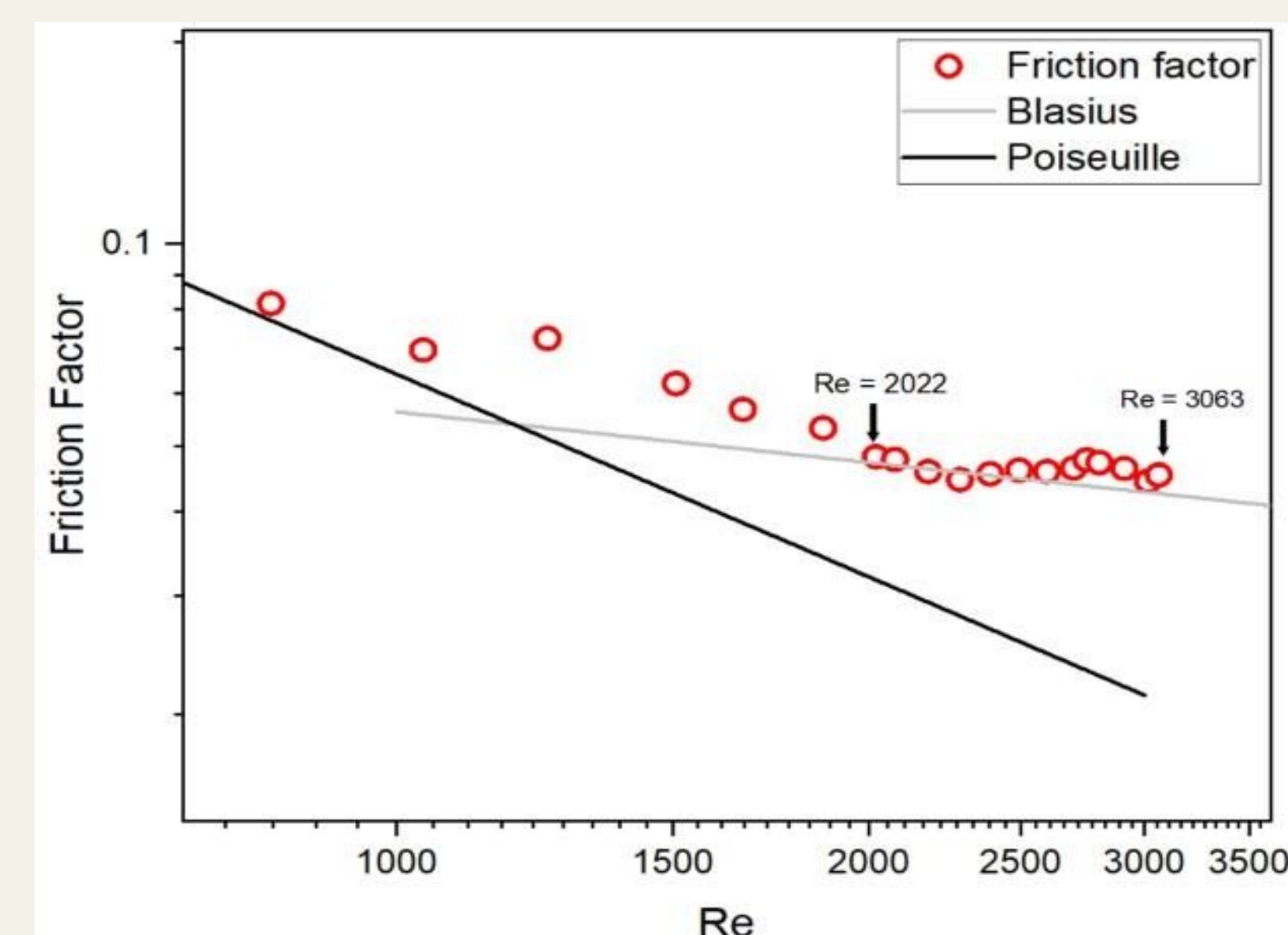
- The friction factor can be calculated by using the differential pressure that was measured, the flux, and the following equation.

$$f = \Delta P \frac{D}{L} \frac{2}{\rho v^2}$$

- When Glyceline is steady at 50°C, 80°C, 100°C, then measure the differential pressure.



- The experiment was carried out until the Reynolds number is 3063. As the figure shows, the value of the differential pressure increases linearly with the Reynolds number.



- As the figure shows, around a Reynolds number of 1000, it looks like the transition region has started. Accordingly, the experimental friction factor value matches with the Poiseuille friction factor and Blasius friction factor. The maximum error of the measured friction factor is 15%.

Conclusion

- The dynamic characteristics of DES, which is considered as an alternative to ionic liquids, were studied as an experimental study.
- The study was conducted by selecting Glyceline as DES and carried out in the laminar and turbulent regions.
- Glyceline was confirmed to be a Newtonian fluid. In the laminar region, the friction factor matched almost the same as the Poiseuille friction factor, and in the turbulent region, it matched almost the same as the Blasius friction factor.
- The fact that Glyceline's dynamic characteristics are much similar to water, known already, therefore, by applying the theoretical values for water, the dynamic characteristics of Glyceline could be judged.

Acknowledgments

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